

CCU-CCUS Initiatives in Pertamina

Presented in CCUS Technical Workshop – ERIA

Dewi Mersitarini, Advisor II-Lead CCUS

Research and Technology Innovation

PT. Pertamina (Persero)

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Pertamina's ESG Strategy to Ensure Environment Sustainability

Priority for Pertamina to define a long term climate ambition and long term energy transition strategy

Top Material Issues and Ambitions



- Addressing Climate Change
 - Reducing Environment Footprint
- Protecting Biodiversity *Deep dive*



- Health & Safety
- Prevention of Major Accidents
- Community Engagement & Impact
- Employee Recruiting, Development & Retention



- Corporate Ethics
- Cyber Security

ESG Strategy across Material Issues

Initiatives	Initiatives to meet the ambitions
Policies	Align policies to achieve and embed ESG practice
Management System	Integrated system to track performance
Disclosures	Sharing achievements with stakeholders

Performance and Results



Better Business Performance



Access to more competitive financing



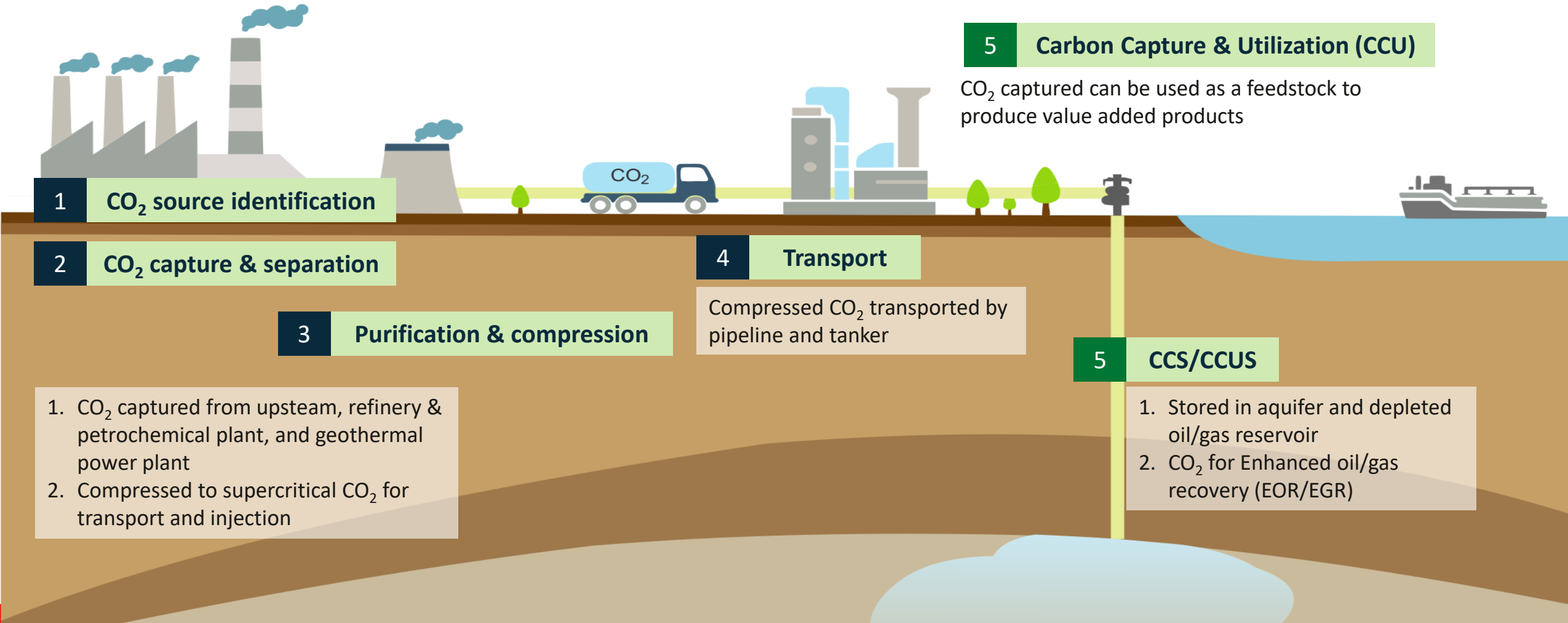
Competitive positioning among peers



Improved ESG scoring by rating agencies

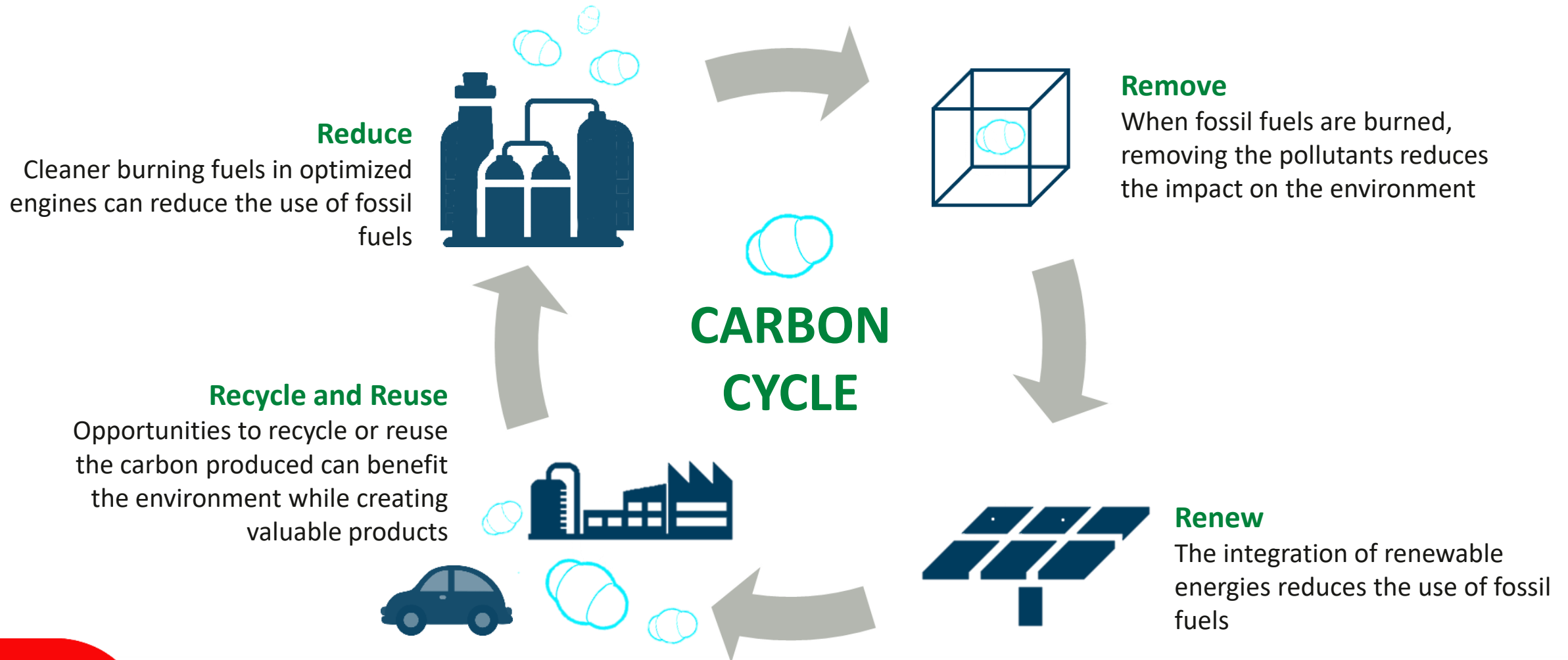
CCS/CCUS Value Chain Development

CCUS is essential to unlock the full potential of decarbonization



Circular Carbon Economy

A circular carbon economy is a framework for managing and reducing emissions. Pertamina have adopted the circular carbon economy framework as a way to reduce their carbon footprints



Potential Project Development

- 1 CCUS CO₂ Enhanced Gas Recovery (EGR)
- 2 CCUS CO₂ Enhanced Oil Recovery (EOR)
- 3 CO₂ Direct Hydrogenation to Methanol
- 4 CO₂ Reforming to Methanol
- 5 CO₂ Mineralization to Precipitated Calcium Carbonate (PCC)
- 6 Blue Hydrogen Production with CCS/CCUS

Research & Development

- 1 CO₂ Mineralization to Prec. Calc. Carbonate (PCC)
- 2 CO₂ Biocapture and Utilization Using Microalgae
- 3 CO₂ Direct Electrolysis to Renewable Methanol
- 4 Catalyst Development for CO₂ Reforming and DME
- 5 Catalyst Development for CO₂ based Polyurethane

Future Potential Breakthrough Project

Large Industrial CCS / CCUS Technology Hubs & Clusters

1. In Situ CO₂ Source and CO₂ Storage Location
2. Cross – boundaries CO₂ Source and CO₂ Storage Location

** Required multi stakeholder collaboration and storage location can be in offshore or onshore*

CCS/CCUS Study and Implementation

Joint Crediting Mechanisme (JCM) scheme covered by Government to Government (G2G) between Indonesia and Japan Government

Gundih Field, Central Java

CO₂ from CPP Gundih will be injected in Kujung formation with potential EGR benefit

Sukowati Field, East Java

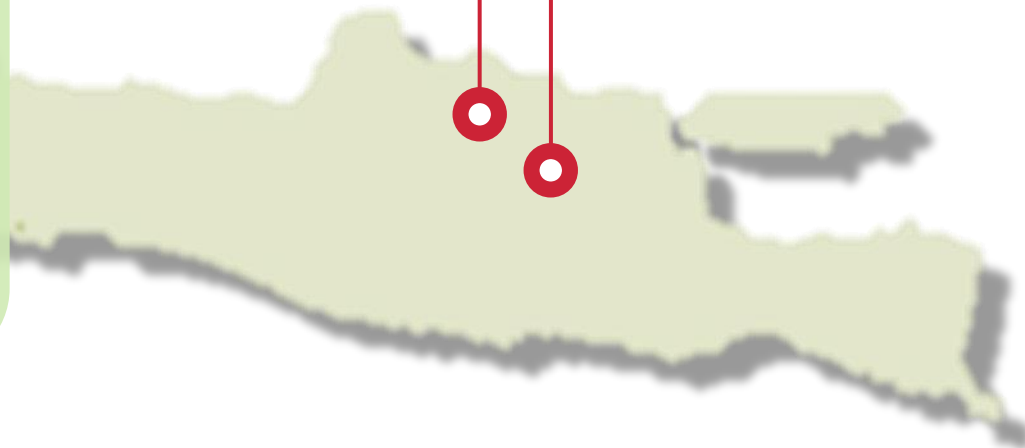
CO₂ from Jambaran Tiung Biru (JTB) will be injected in Tuban layer with EOR benefit

Potential of CO₂ Reduction from CCUS Process in Upstream Field:

18 Million ton CO₂

Gundih: 3 Million ton CO₂ for 10 years
&

Sukowati: 15 Million ton CO₂ for 25 years



The utilization of CO₂ has the potential to benefit carbon reduction

- Potential enhanced hydrocarbon production
- Stored CO₂ will be monetized in Carbon market, after fulfill NDC contribution

CO₂ Utilization

Carbon utilization is the use of CO₂ to create products with economic value

Chemicals

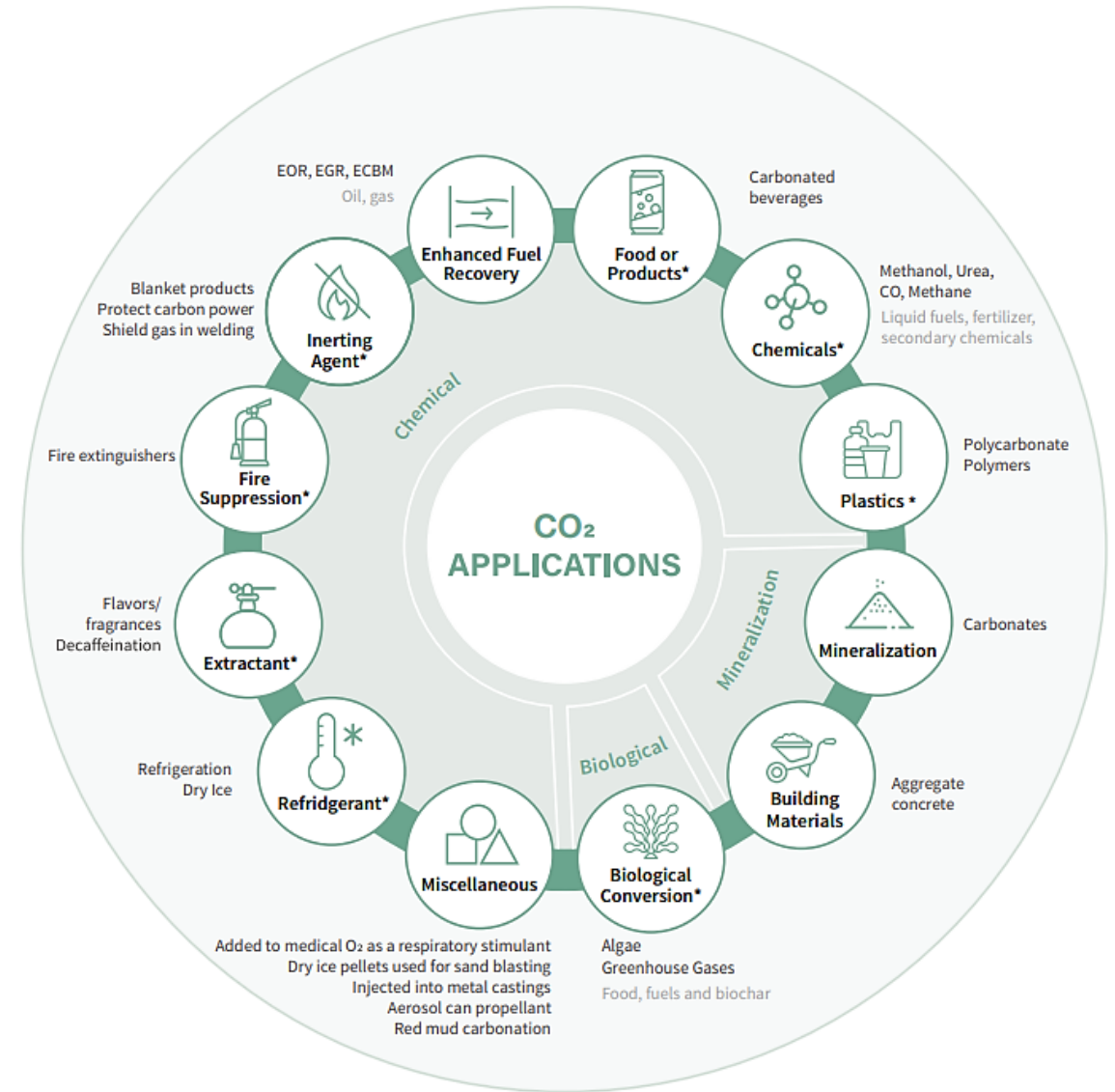
CO₂ is currently used in small quantities to make urea fertiliser and some special polymers. In a future hydrogen economy, CO₂ could be combined with H₂ to make synthetic fuels, syngas and methanol. Syngas and methanol are basic chemical feedstock from which many chemicals and polymers can be made

Biological

CO₂ is used to promote plant growth and can be captured in soils by using biochar to increase soil quality

Mineralization

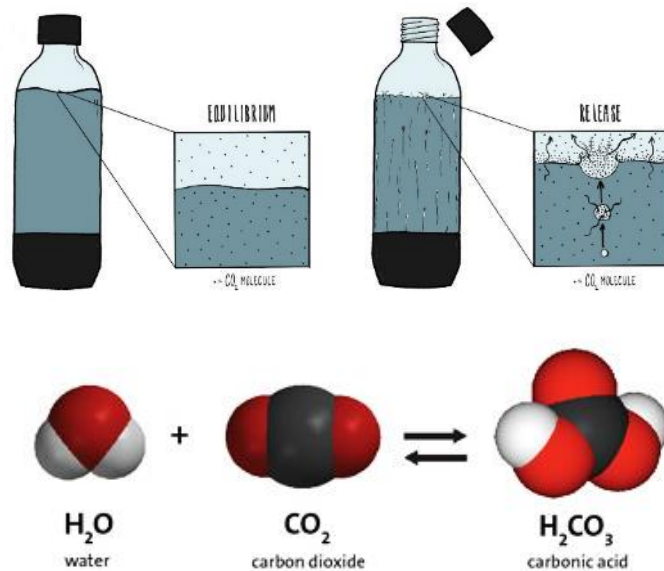
Incorporating CO₂ into concrete has the most potential to become a large market for CO₂ in the near term. This process is energy efficient using minimal external energy



Conventional CO₂ Utilization

Carbonate Drinks

The water will have absorbed as much CO₂ as it can. Chemists call this an equilibrium. Fizzy drinks do release carbon dioxide (CO₂), a can of pop contains 2-3 g of CO₂



Carbonic acid (H₂CO₃) also has a mild antibiotic effect that prevents bacteria from growing in the water

Dry Ice (CO₂) as cooling agent

The carbon dioxide-rich gas is pressurized and refrigerated until it liquefies. Next, the pressure is reduced. As a result, the extreme cold causes the liquid to solidify into a snow-like consistency. Finally, the snow-like solid carbon dioxide is compressed into small pellets or larger blocks of dry ice ^[a]



HOW DRY ICE IS MADE



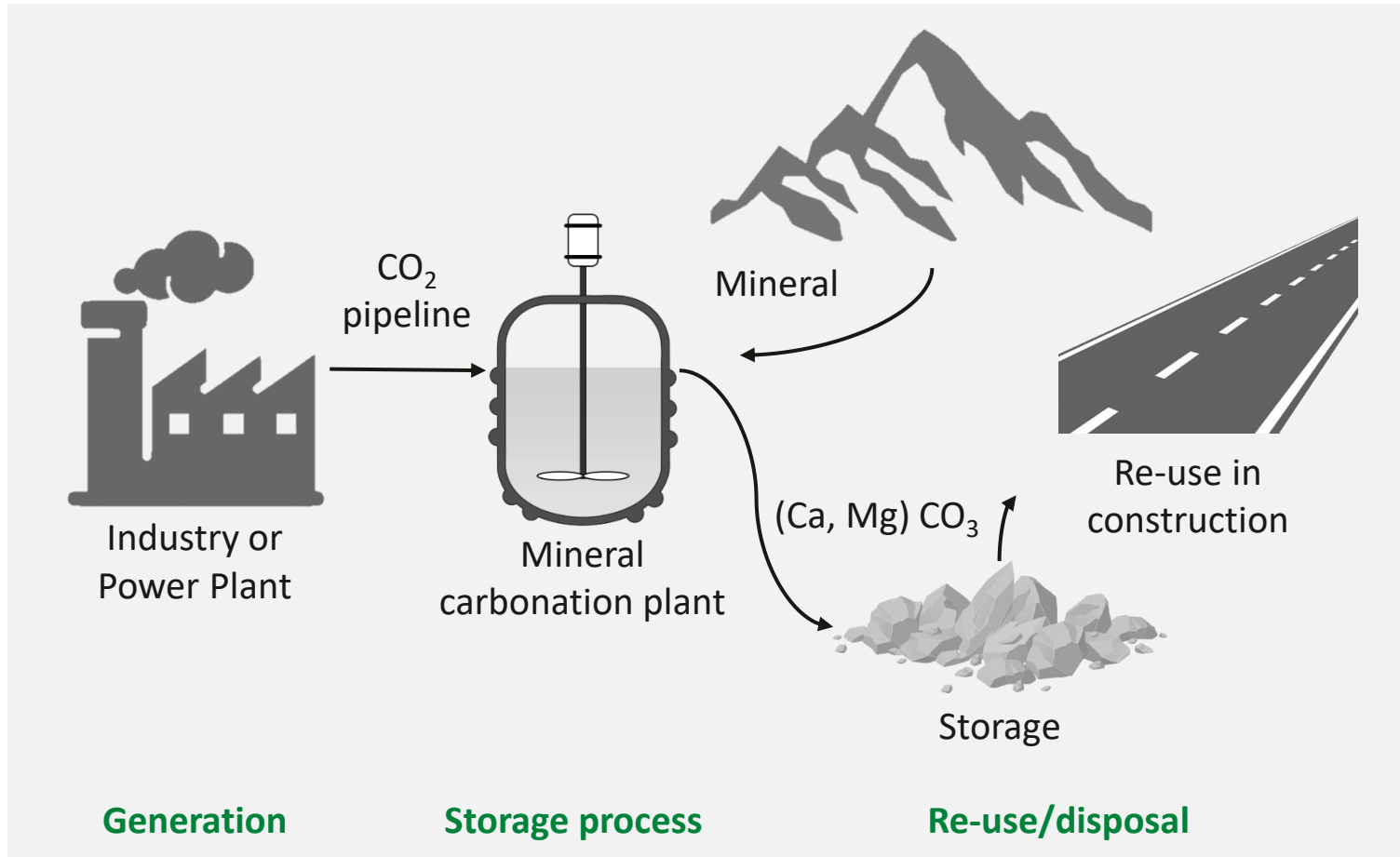
Source:

[a] <https://www.cryonomic.com/en/dry-ice-process>

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CO₂ to Mineralization

Mineralization of carbon dioxide (CO₂), or mineral carbonation, involves the reaction of CO₂ with materials containing alkaline-earth oxides like magnesium oxide (MgO) and calcium oxide (CaO)



Mineral Carbonation Reaction:



(Where M is a divalent metal, e.g. calcium, magnesium, or iron)

Mineral Carbonation Uses

Precipitated Calcium Carbonate (PCC):

- Paper Industry
- Paint Industry
- Tire Industry
- Pharmacy Industry
- Cosmetics Industry
- Food Industry
- Potential as a PCC technology licensor

CO₂ Capture and Utilization to Precipitated Calcium Carbonate (PCC)

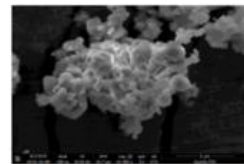
- **Background :** CO₂ utilization through mineralization route requires much lower energy compare to other method
- **Research Objective :** Reducing CO₂ emission while utilizing it and creating efficient PCC production process



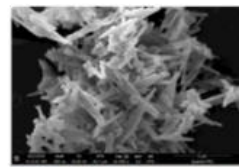
PCC Production Process: $CaO + NH_4Cl + CO_2 \rightarrow CaCO_3 + NH_4Cl$



Precipitated Calcium Carbonate (PCC)



PCC with Calcite Structure



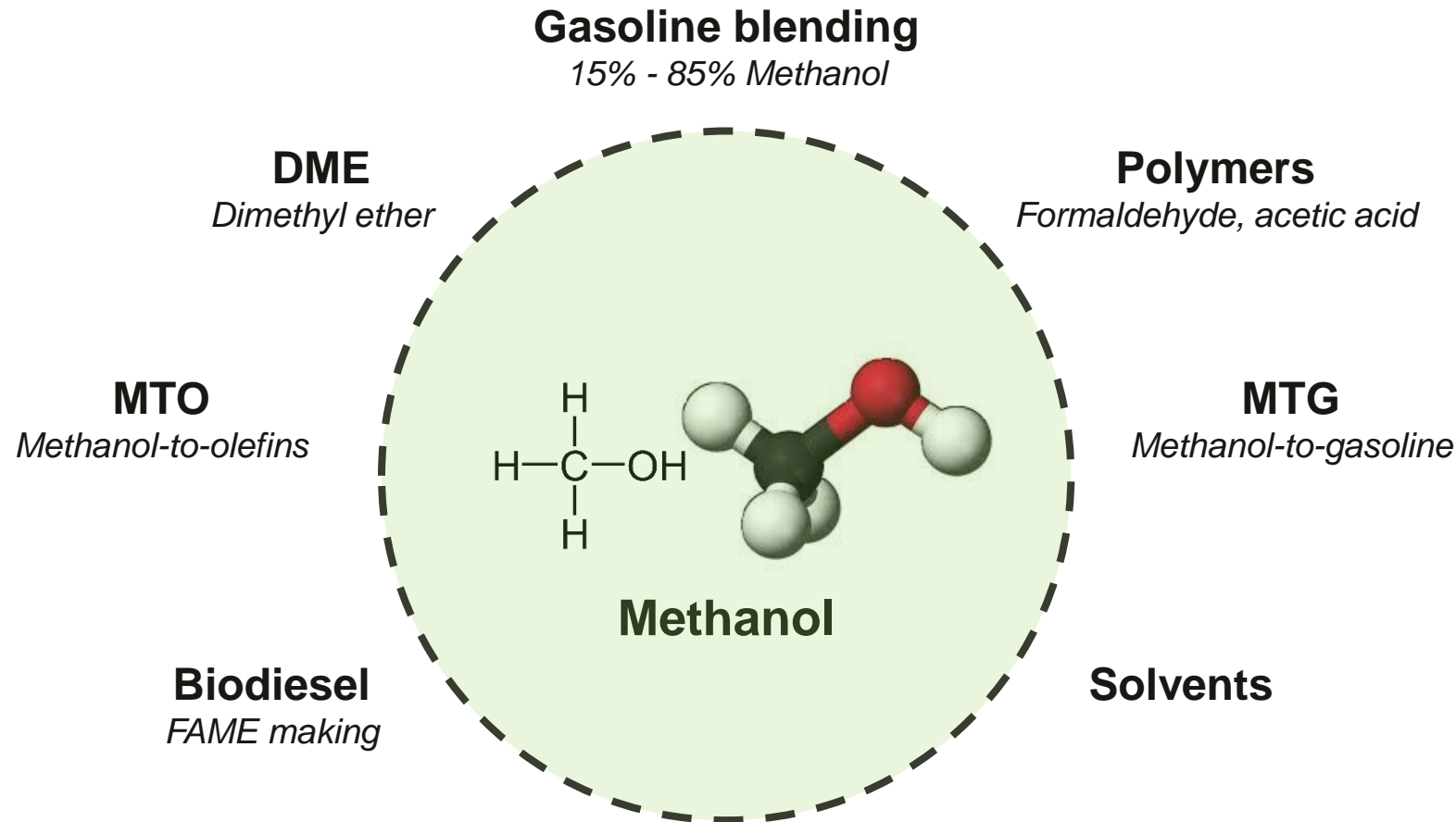
PCC with Aragonite Structure

Achievements:

- High purity CaCO₃ >99%
- CO₂ conversion up to 70%
- NH₄Cl as a solvent to increase CaO solubility
- Aragonite structure PCC product for high grade

Uses of Methanol

Methanol occupies a critical position in the chemical industry, as a highly versatile building block for the manufacture of countless everyday products



Methanol is one of the four critical basic chemicals – alongside ethylene, propylene and ammonia – used to produce all other chemical products

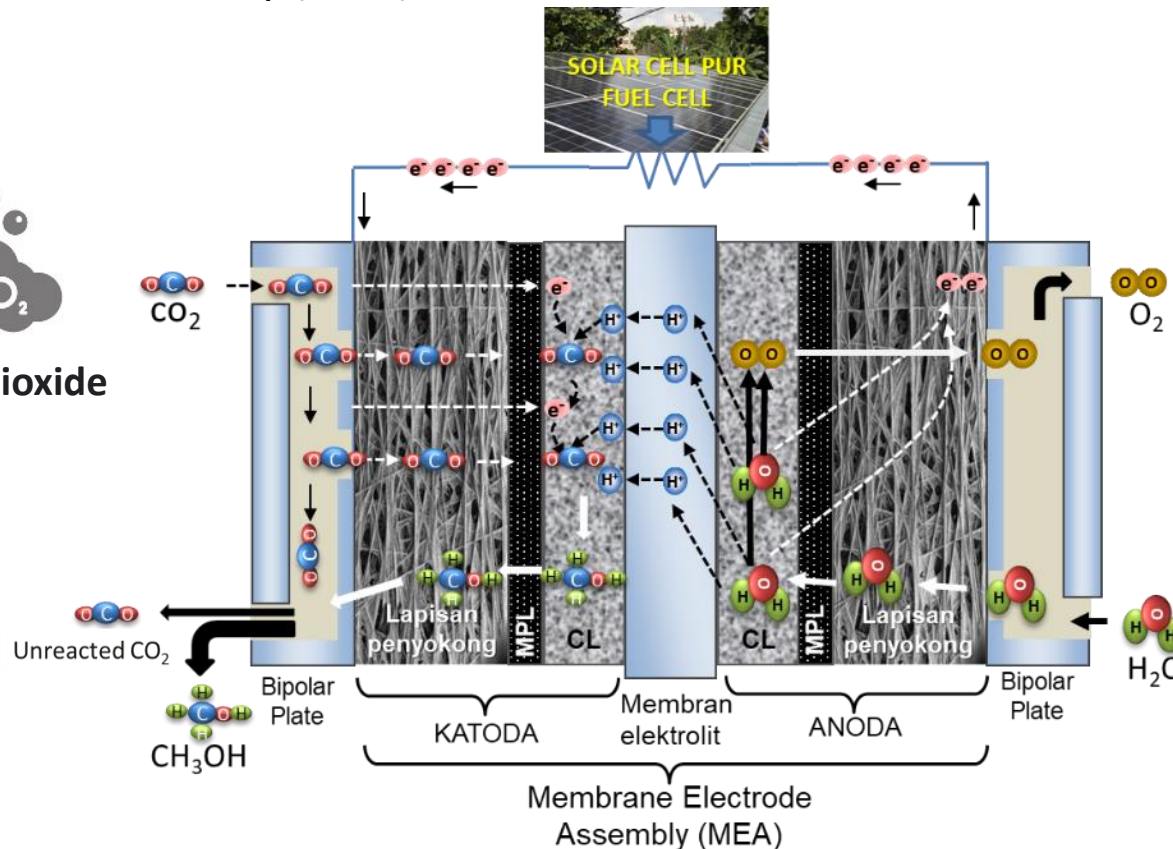
CO₂ Capture and Utilization to Renewable Methanol

Power to Liquid (PtL) to convert renewable energy to liquid fuels and chemicals using Membrane Electrode Assembly (MEA)


Carbon dioxide



Renewable Methanol
> 30% concentration



Renewable Energy source for
electrolysis

Electricity from renewable sources is available, hydrogen is produced from water electrolysis

Hydrogen combined with carbon dioxide then converted into renewable methanol

CO₂ Capture and Utilization to Methanol

Pertamina Business Initiative in developing CO₂ based chemicals and fuels through applied technology implementation

Geothermal Field

Green Methanol

- **CO₂ sourced** from non-condensable gas as a content of steam
- **H₂ produced** by electrolysis of water using renewable electricity

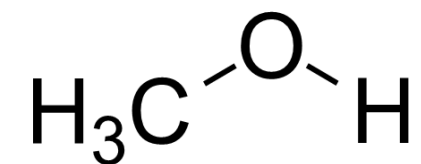
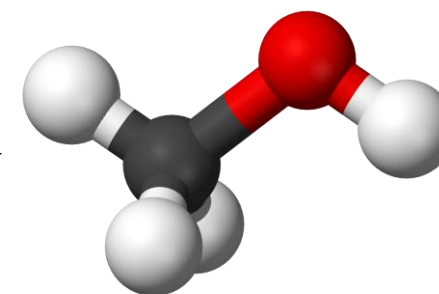


Upstream & Refinery

Blue Methanol

1. **CO₂ reforming of methane**
CO₂ and CH₄ sourced from upstream
$$\text{CO}_2 + \text{CH}_4 + \text{low steam} \rightarrow \text{Syngas} \rightarrow \text{CH}_3\text{OH}$$
2. **CO₂ from refinery field and H₂ produced by electrolysis of water** using renewable electricity or hydrogen recovery

Methanol

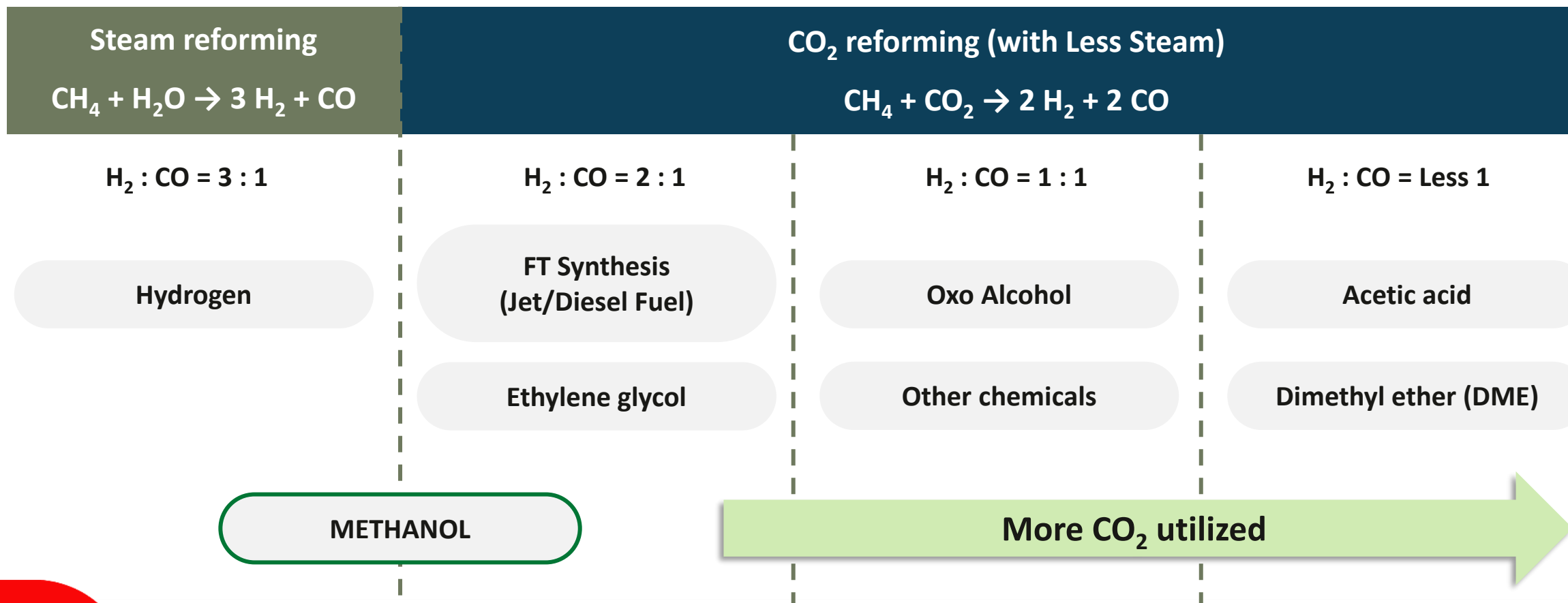


Methanol for gasoline blending (Up to 15% methanol) & Support FAME Industry

Natural gas with High CO₂ Utilization

CO₂ as feedstock integrated with NG conversion process (NG-CO₂ reforming) to produce high value-added products

Steam reforming of methane combined with CO₂ or commonly referred to as **CO₂ reforming of methane**



Catalyst Development for CO₂ Reforming & Direct DME

Catalyst development based of Cobalt – Nickel (Co-Ni) and Copper (Cu) to produce DME from natural gas

UPSTREAM GAS FIELD

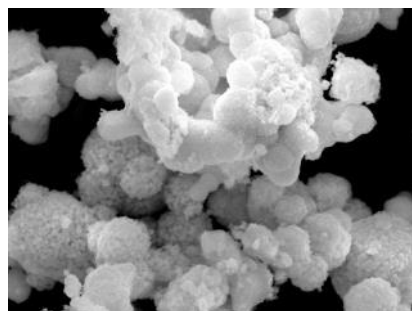
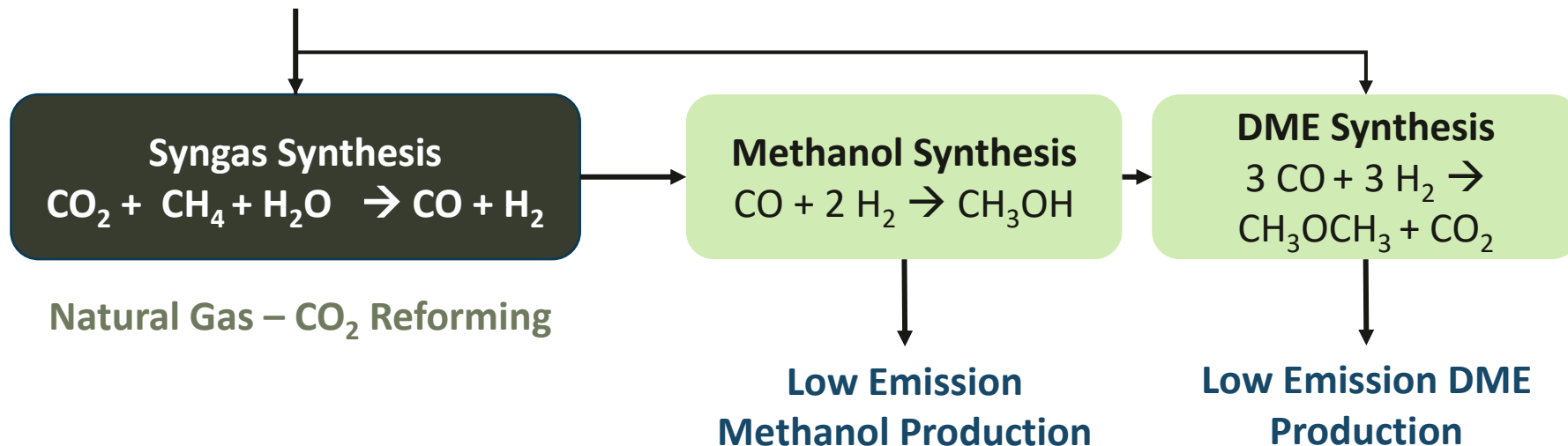


Carbon dioxide
up to 30% of CO₂ content in
Natural Gas



Additional Steam

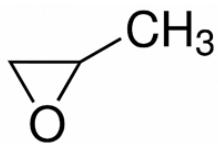
Direct Dimethyl Ether (DME) Synthesis from Syngas



- Gasoline Blending (A20 program)
- LPG Blending
- Chemical Industry
- FAME Industry
- RTI Internal Research Usage (DMFC Fuel Cell, Direct Methanol Fuel)

CO₂ based Polyols for Polyurethane Synthesis

Polyurethane made from CO₂-based polyol and di-isocyanate



Propylene oxide
(PO)

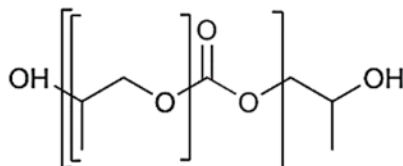
&

+ CO₂



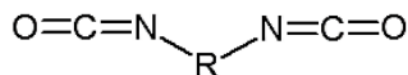
Ethylene oxide
(EO)

Containing up to
40%-wt CO₂

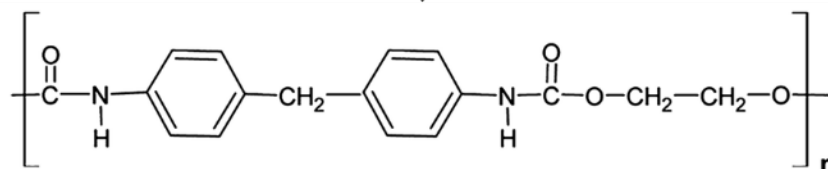


CO₂-based Polyol

+



Di-isocyanate



Polyurethane

Coatings – a high degree of hardness while

maintaining sufficient flexibility

Coil, can, automotive, plastic, wood, textile or leather coatings



Adhesives & sealants

Flexible packaging adhesives, reactive hot melts, structural adhesives, and used in many applications



Elastomers

Thermoplastic polyurethane

Toys, medical application, automotive, etc.

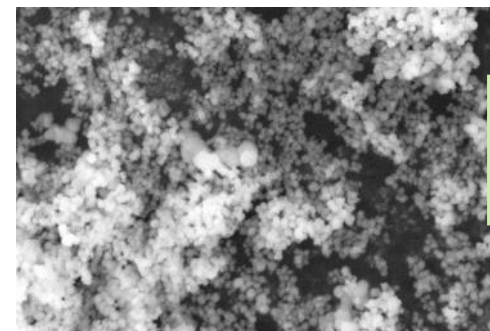
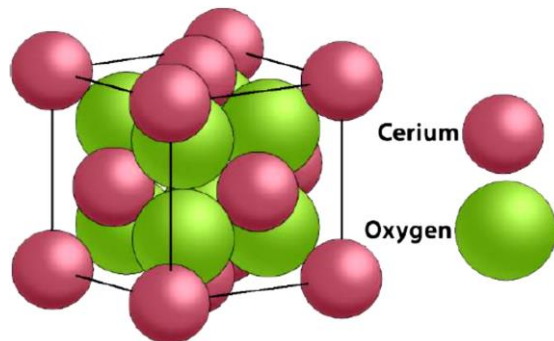
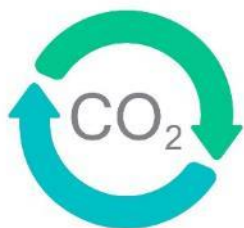
Polyols synthesis by reaction of CO₂ with epoxides (ethylene oxide and propylene oxide)

CO₂-based Polyol as an intermediate product
for Polyurethane synthesis

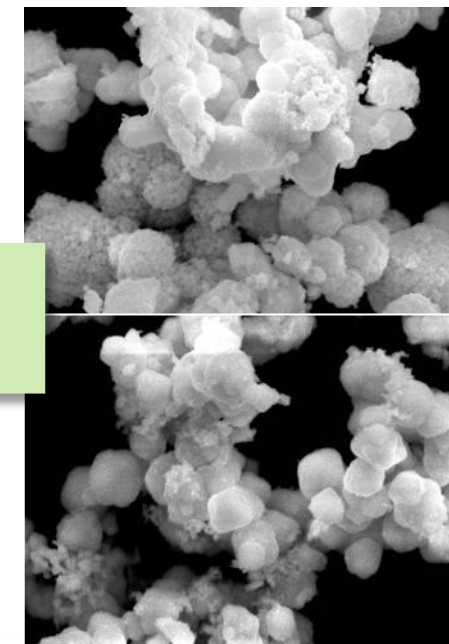
Catalyst Development for CO₂ Utilization to Polymer

By reaction of non-isocyanate polyurethane (NIPU) using CO₂ and alcohol as feedstock

Catalyst CeO₂ with other transition metals doping to enhance the activity of the catalyst



Catalyst CeO₂ without metal doping



Catalyst CeO₂ with metal doping

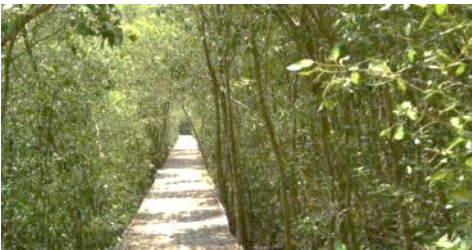
The particle catalyst size increases 2 - 5 times with addition of other transition metals

CO₂ Biocapture

Algae and forests have great potential in the reduction of CO₂ emissions

Planted species	Removal rate
Forest (<i>Eucalyptus</i>)	38.0 – 40.7 t CO ₂ /ha/year
Forest (<i>Pine</i>)	10.2 – 21.1 t CO ₂ /ha/year
Mangrove tree restoration	Up to 23.1 t CO ₂ /ha/year ^a
Algae dry material (<i>C. vulgaris</i>)	1.6 – 2 kg CO ₂ /kg/year ^b

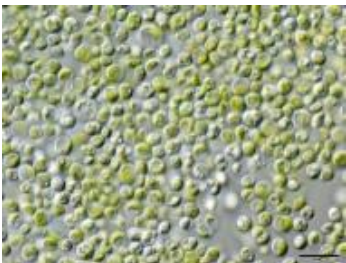
For forest during the first 20 years of growth



Planting woodland to remove CO₂ from the atmosphere, known as woodland carbon capture, **is a cost-effective way of compensating for emissions** while also providing many other social and environmental benefits

Growth Parameters for Microalgae

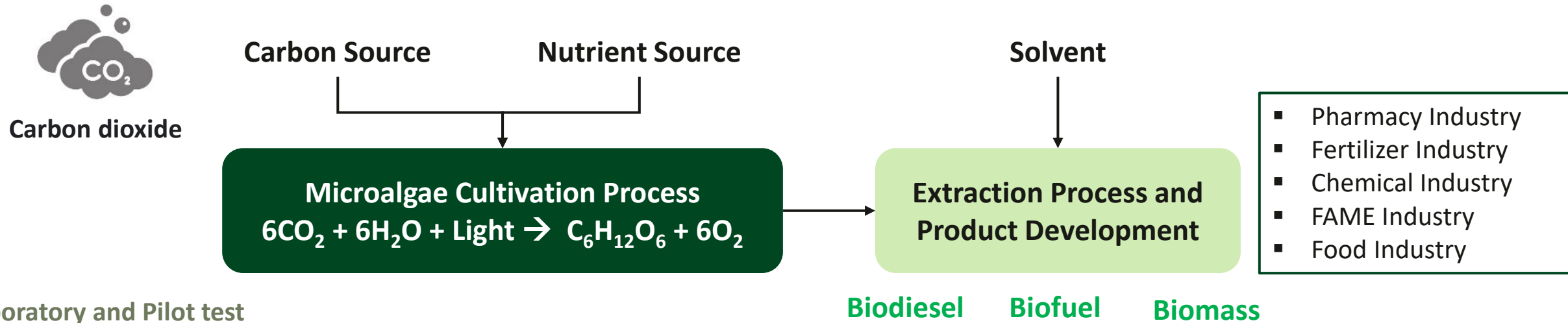
Species	CO ₂ (%)	Features ^c
<i>Chlorococcum sp</i>	8	High CO ₂ fixation rate Densely culturable
<i>Chorella sp</i>	Up to 60	High growth ability, High temperature tolerance
<i>Euglena gracilis</i>	Up to 100	High amino acid content, Good digestibility (effective fodder) Grows well under acidic conditions, not easily contaminated
<i>Galdieria sp</i>	Up to 100	High CO ₂ tolerance
<i>Synechococcus lividus</i>	Up to 70	High pH tolerance



Source:
[a] Bernal et al. (2018): Global carbon dioxide removal rates from forest landscape restoration activities
[b] Carlsson et al. : Micro-and Macro-Algae: Utility for Industrial Applications
[c] The Comprehensive Guide for Algae – based Carbon Capture (www.oigae.com)

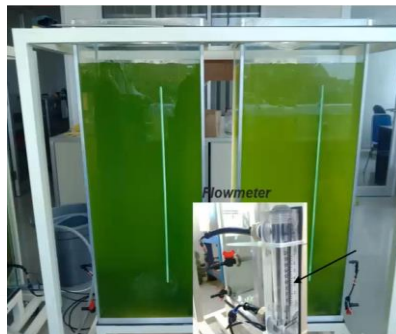
CO₂ Biocapture and Utilization using Microalgae

Specific algae strains will be tested in CO₂ concentrated carbon source



Laboratory and Pilot test
Development for Field Test

[Lab. Scale PBR – 100 L]



[Pilot Vertical PBR Reactor
– 1,000 L]



Economic commercial development
with Open Pond cultivation

[Commercial Open Pond – 20,000 L]



Key Takeaways Development for Decarbonization

Robust legal and regulatory frameworks are needed to facilitate the development of CCUS to provide project developers and financiers with certainty and confidence to invest in green and low carbon projects. They will also play a key role in ensuring the protection of public health and the environment

	Support from the government	Collaboration and partnership opportunity
EXISTING	<ul style="list-style-type: none">▪ Research funding/grant by Indonesia Endowment Fund for Education (LPDP) and Ministry of Education and Culture▪ Super tax deduction for decarbonization research by Ministry of Finance	<ul style="list-style-type: none">▪ Green financing and softloan▪ Partnership to share project cost and benefit, sharing risk and to build required technology capability▪ Carbon trading program▪ Technology improvement and optimization to reduce Capex & Opex▪ Creating CCUS market and stakeholders acceptance
FUTURE	<ul style="list-style-type: none">▪ Carbon Economy Value (NEK) by Ministry of Environment and Forestry▪ Regulations for implementing CCS/CCUS by Ministry of Energy and Mineral Resources	

The appropriate policy mix will depend on local market conditions and institutional factors, as well as the stage of technology development in the sector or application

THANK YOU

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Ketulusan untuk Melayani